

- FURNAS, T. C. (1957). *Single Crystal Orientor Instruction Manual*. Milwaukee: General Electric Company.
- HANSON, H. P., HERMAN, F., LEA, J. D. & SKILLMAN, S. (1964). *Acta Cryst.* **17**, 1040.
- HOARD, J. L. & SILVERTON, J. V. (1963). *Inorg. Chem.* **2**, 235.
- International Tables for X-ray Crystallography* (1952). Vol. I. Birmingham: Kynoch Press.
- International Tables for X-ray Crystallography* (1962). Vol. III. Birmingham: Kynoch Press.
- KARLE, J. & KARLE, I. (1965). In *Computing Methods in X-ray Crystallography*. J.S. ROLLETT, editor. Oxford: Pergamon Press.
- MOONEY, R. C. L. (1949). *Acta Cryst.* **2**, 189.
- PAULING, L. (1960). *The Nature of the Chemical Bond*. Ithaca: Cornell Univ. Press.
- PEARSON, W. B. (1958). *A Handbook of Lattice Spacings and Structures of Metals and Alloys*. New York, London, Paris, Los Angeles: Pergamon Press.
- SMITH, G. S., JOHNSON, Q. & NORDINE, P. C. (1965). *Acta Cryst.* **19**, 668.
- STEIN, L. (1964). *Inorg. Chem.* **3**, 995.
- WYCKOFF, R. G. W. (1963). *Crystal Structures*. Vol. 1, 2nd edition. New York, London, Sydney: Interscience.
- ZACHARIASEN, W. H. (1954). In *The Actinide Elements*. G.T. SEABORG & J.J. KATZ, editors. New York, Toronto, London: McGraw-Hill.

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The Crystal and Molecular Structure of 3β -Acetoxy- 17α -hydroxy- 16β -bromo- 5α -pregnan-11,20-dione*

BY JEAN M. OHRT, BARBARA A. HANER,[†] A. COOPER,[‡] AND DORITA A. NORTON[‡]

Center for Crystallographic Research, Roswell Park Memorial Institute, Buffalo, New York

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The crystal structure of 3β -acetoxy- 17α -hydroxy- 16β -bromo- 5α -pregnan-11,20-dione has been determined by three-dimensional X-ray analysis and refined to a final *R* value of 0.090. Unit cell constants are $a=11.521$, $b=19.587$, $c=10.390 \text{ \AA}$, space group $P2_12_12_1$. Standard deviations for bond lengths and angles are in the ranges of 0.02–0.03 \AA and $1\text{--}2^\circ$ respectively. Neighboring molecules are hydrogen bonded via the 11-ketone and 17-hydroxyl groups. Molecular geometry and packing of the molecules are discussed.

Introduction

Many hypotheses may be formulated in an attempt to explain the molecular stacking arrangements of steroids in the crystalline state, such as the presence or absence of heavy atoms, angular methyl groups, side chains, etc. (Norton, 1965; Norton & Ohrt, 1966). The present paper presents the first in a series of structure determinations undertaken to augment the information presently available in this area. Three steroids of the 5α -pregnane series have been selected. Two, 3β , 17α -dihydroxy- 21 -bromo- 5α -pregnan-11,20-dione and 3β , 17α -dihydroxy- 16β -bromo- 5α -pregnan-11,20-dione, are isomers, the difference being in the position of attachment of the heavy atom; the third varies in that it has an acetate side chain attached to carbon 3, namely, 3β -acetoxy- 17α -hydroxy- 16β -bromo- 5α -pregnan-11,20-dione (Fig. 1), the structure of which is herein presented.

Experimental

Crystals of the steroid were grown from methanol and from 95% ethanol solutions by slow evaporation at room temperature. These solvents provided two different crystal modifications: monoclinic from methanol and orthorhombic from ethanol. Crystal data for the two modifications are:

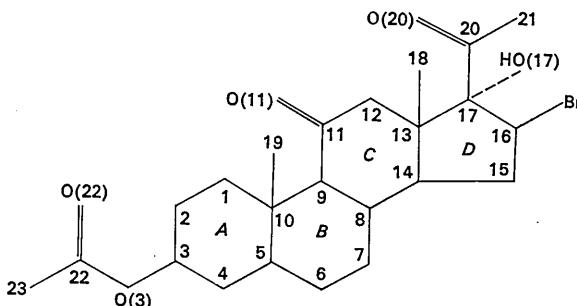


Fig. 1. 3β -Acetoxy- 17α -hydroxy- 16β -bromo- 5α -pregnan-11,20-dione. Schematic drawing showing the numbering scheme used in the discussion.

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[†] Deceased 11 February 1966.

[‡] Present address: The Medical foundation of Buffalo, 73 High Street, Buffalo, New York, U.S.A.

Monoclinic modification

Solvent methanol, $a = 7.142 \pm 0.001$, $b = 26.208 \pm 0.008$, $c = 7.088 \pm 0.001$ Å, $\beta = 113.56 \pm 0.02^\circ$, $V_0 = 1216.18$ Å³, $D_m = 1.292 \pm 0.005$ g.cm⁻³, $Z = 2$, $D_c = 1.274$ g.cm⁻³. Space group $P2_1$ (C_2^2 ; no. 4). Thin plates.

Orthorhombic modification

Solvent ethanol, $a = 11.521 \pm 0.003$, $b = 19.587 \pm 0.001$, $c = 10.390 \pm 0.001$ Å, $V_0 = 2344.63$ Å³, $D_m = 1.314 \pm 0.005$ g.cm⁻³, $Z = 4$, $D_c = 1.321$ g.cm⁻³. Space group $P2_12_12_1$ (D_2^4 ; no. 19). Prismatic blocks.

Owing to the more regular nature of the orthorhombic crystals, these were used for the intensity measurements, for which a crystal of dimensions $0.3 \times 0.1 \times 0.05$ mm was selected. Intensity measurements were made with a General Electric single-crystal orienter by the stationary-counter-stationary-crystal method using balanced nickel and cobalt filters (ten seconds count with each filter). A total of 2396 reflections were measured with the use of $Cu K\alpha$ radiation. Approximate absorption corrections were applied and the intensities were corrected for Lorentz and polarization effects.

A three-dimensional Patterson synthesis was computed from $|F|^2$ values modified to represent point atoms with thermal motion, and the bromine atom was located from the Harker sections. The positions of the other twenty-eight carbon and oxygen atoms in the asymmetric unit were determined by routine application of the heavy atom method. No attempt was made to locate the hydrogen atoms.

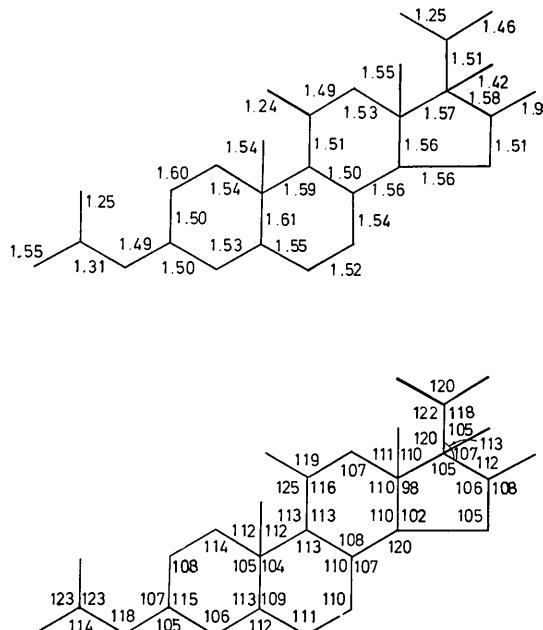


Fig. 2. Interatomic distances and angles. Standard deviations are in the ranges of 0.02 and 0.03 Å and 1–2° respectively.

The positional and thermal parameters of all atoms and the anisotropic thermal parameters of the bromine atom were refined by block-diagonal least squares in which the bromine atom and C(16), to which the bromine atom is bonded, were included in a single enlarged (13 × 13) matrix block to take account of interaction between these two atoms. The refinement converged with an R index (defined as $\sum ||F_o|| - |F_c|| / |F_o||$) of 0.090 omitting 'unobserved' reflections. During the refinement, the atomic scattering factors used were those given in *International Tables for X-ray Crystallography* (1962) and the bromine scattering factors were corrected for the real part of the anomalous dispersion correction only. The quantity minimized was $\sum w(|F_o| - |F_c|)^2$ where $w^{1/2} = 1$ if $|F_o| \leq 20$ and $w^{1/2} = 20/|F_o|$ if $|F_o| > 20$. The refined parameters are listed in Table 1.

Table 1. Positional and vibrational parameters of the atoms

The standard deviations in the last two digits of each number are given in parenthesis.

	x/a	y/b	z/c	$B(\text{iso})$
Br	0.5236 (2)	0.3261 (1)	1.0087 (3)	6.32 (-)*
O(3)	0.4402 (11)	0.9264 (6)	1.0303 (12)	4.72 (27)
O(11)	0.3520 (12)	0.6323 (7)	0.7226 (13)	5.25 (30)
O(17)	0.2183 (10)	0.4350 (6)	1.0094 (14)	4.61 (25)
O(20)	0.3159 (13)	0.3731 (7)	0.7383 (14)	5.32 (30)
O(22)	0.2556 (17)	0.9607 (10)	1.0128 (21)	9.64 (49)
C(1)	0.3472 (18)	0.7581 (9)	0.8765 (18)	4.54 (42)
C(2)	0.3780 (17)	0.8375 (9)	0.8851 (18)	4.25 (40)
C(3)	0.4013 (16)	0.8538 (9)	1.0241 (19)	4.58 (39)
C(4)	0.4979 (17)	0.8133 (9)	1.0839 (19)	4.34 (41)
C(5)	0.4584 (15)	0.7388 (8)	1.0799 (17)	3.52 (34)
C(6)	0.5438 (21)	0.6909 (11)	1.1512 (23)	5.93 (52)
C(7)	0.4965 (17)	0.6184 (9)	1.1600 (20)	4.57 (41)
C(8)	0.4733 (15)	0.5900 (8)	1.0245 (17)	3.66 (32)
C(9)	0.3912 (14)	0.6361 (7)	0.9530 (15)	2.74 (31)
C(10)	0.4403 (16)	0.7111 (8)	0.9351 (17)	3.58 (36)
C(11)	0.3455 (17)	0.6052 (9)	0.8297 (19)	4.17 (39)
C(12)	0.2908 (15)	0.5363 (8)	0.8408 (16)	3.16 (33)
C(13)	0.3795 (16)	0.4896 (9)	0.9065 (17)	3.73 (37)
C(14)	0.4121 (14)	0.5196 (7)	1.0410 (14)	2.92 (32)
C(15)	0.4770 (19)	0.4586 (10)	1.1057 (18)	5.00 (43)
C(16)	0.4107 (19)	0.3959 (10)	1.0657 (20)	4.83 (42)
C(17)	0.3278 (15)	0.4198 (7)	0.9529 (15)	2.88 (33)
C(18)	0.4862 (18)	0.4774 (9)	0.8193 (18)	4.57 (39)
C(19)	0.5532 (18)	0.7122 (10)	0.8563 (19)	4.54 (42)
C(20)	0.3033 (18)	0.3639 (9)	0.8568 (19)	4.44 (41)
C(21)	0.2516 (20)	0.3004 (11)	0.9034 (23)	5.70 (51)
C(22)	0.3608 (19)	0.9738 (10)	1.0249 (21)	5.76 (47)
C(23)	0.4112 (20)	1.0474 (11)	1.0196 (23)	5.89 (47)

* Equivalent $B(\text{iso})$. The anisotropic thermal parameters for the bromine atom in the form $\exp[-\sum(h_i h_j b_{ij} \times 10^{-5})]$ are:

$$\begin{array}{cccccc} b_{11} & b_{22} & b_{33} & b_{12} & b_{31} & b_{23} \\ 1176 (19) & 369 (6) & 1635 (28) & 262 (10) & -44 (28) & 38 (15) \end{array}$$

Discussion

The least-squares planes through important regions of the steroid nucleus, distances of atoms from these planes and interplanar angles are given in Table 2 and the interatomic distances and angles are summarized in Fig. 2. The least-squares planes through rings A , B , C , and D are almost parallel [Table 2(c)] and the r.m.s.

Table 2(a). Least-squares planes through the atoms*

Plane	Atoms	<i>l</i>	<i>m</i>	<i>n</i>	<i>p</i>
<i>A</i> 1	C(2), C(3), C(4)	0.6558	0.7021	-0.2774	11.823
<i>A</i> 2	C(1), C(2), C(4), C(5)	0.8093	-0.1920	-0.5551	-4.700
<i>A</i> 3 <i>B</i> 1	C(1), C(5), C(6), C(10)	0.7406	0.5942	-0.3138	8.956
<i>B</i> 2	C(6), C(7), C(9), C(10)	0.7944	-0.3529	-0.4943	-5.698
<i>B</i> 3 <i>C</i> 1	C(7), C(8), C(9), C(11)	0.7724	0.4709	-0.4263	5.051
<i>C</i> 2	C(8), C(11), C(12), C(14)	0.7167	-0.4106	-0.5637	-6.855
<i>C</i> 3 <i>D</i> 1	C(12), C(13), C(14), C(15)	0.8164	0.4067	-0.4100	3.512
<i>D</i> 2	C(13), C(15), C(16), C(17)	0.7205	-0.3944	-0.5703	-6.067
<i>D</i> 3	C(14), C(15), C(16), C(17)	-0.7020	0.0681	0.7090	4.977
<i>A</i>	C(1), C(2), C(3), C(4), C(5), C(10)	0.8487	0.1265	-0.5135	0.825
<i>B</i>	C(5), C(6), C(7), C(8), C(9), C(10)	0.8775	-0.0293	-0.4787	-0.884
<i>C</i>	C(8), C(9), C(11), C(12), C(13), C(14)	0.8055	-0.1069	-0.5829	-3.281
<i>D</i>	C(13), C(14), C(15), C(16), C(17)	0.8061	-0.1925	-0.5596	-3.891
C(1)-C(17)		0.8275	-0.0550	-0.5587	-2.486

* The planes are in the form $lx+my+nz=p$, where x, y, z and p are in Å and x, y and z are referred to the crystal axes with $x||a$, $y||b$ and $z||c$.

Table 2(b). Distances from the least-squares planes

	<i>A</i>	<i>B</i>	<i>C</i>	<i>D</i>	<i>A</i> 2	<i>A</i> 3 <i>B</i> 1	<i>B</i> 2	<i>B</i> 3 <i>C</i> 1	<i>C</i> 2	<i>C</i> 3 <i>D</i> 1	<i>D</i> 2	<i>D</i> 3	C(1)-C(17)
C(1)	-0.23					0.03	-0.03						-0.11
C(2)	0.22						-0.03						0.05
C(3)	-0.25						-0.68						-0.55
C(4)	0.27						0.03						0.07
C(5)	-0.27	-0.28				-0.03	0.03	-0.76					-0.21
C(6)		0.26					-0.03	-0.01					0.24
C(7)		-0.22						0.01	-0.07				-0.18
C(8)		0.23	0.23					0.69	-0.06	0.02			0.42
C(9)		-0.27	-0.19					-0.01	0.08	-0.61			-0.00
C(10)	0.25	0.28				0.74	0.03	0.01					0.49
C(11)			0.19						-0.08	-0.02			0.31
C(12)			-0.24							0.02	-0.09		-0.20
C(13)			0.29	0.30						0.74	0.10	0.06	-0.72
C(14)			-0.29	-0.29						-0.02	0.07	0.69	0.32
C(15)				0.16							-0.08	-0.07	-0.08
C(16)				0.02								0.10	0.08
C(17)				-0.19								-0.10	-0.05

deviation of atoms C(1)-C(17) from the least-squares plane through the steroid nucleus is 0.28 Å. The length of the steroid nucleus [C(3)-C(16)] is 8.98 Å. Rings *A*, *B*, *C*, and *D* are *trans*-fused and rings *A*, *B*, and *C* are chair shaped. Ring *D* has a distorted half-chair configuration with torsional angles C(14)-C(15)-C(16)-C(17) and C(15)-C(16)-C(17)-C(13) of +13° and +17° respectively.

Table 2(c). Interplanar angles

Plane 1	Plane 2	Angle
<i>A</i> 1	<i>A</i> 2	123°
<i>A</i> 3 <i>B</i> 1	<i>A</i> 2	131
<i>A</i> 3 <i>B</i> 1	<i>B</i> 2	122
<i>B</i> 3 <i>C</i> 1	<i>B</i> 2	131
<i>B</i> 3 <i>C</i> 1	<i>C</i> 2	127
<i>C</i> 3 <i>D</i> 1	<i>C</i> 2	130
<i>C</i> 3 <i>D</i> 1	<i>D</i> 2	131
<i>C</i> 3 <i>D</i> 1	<i>D</i> 3	33
<i>A</i>	<i>B</i>	171
<i>B</i>	<i>C</i>	171
<i>C</i>	<i>D</i>	175
C(1)-C(17)	(100)	146
C(1)-C(17)	(010)	87
C(1)-C(17)	(001)	56

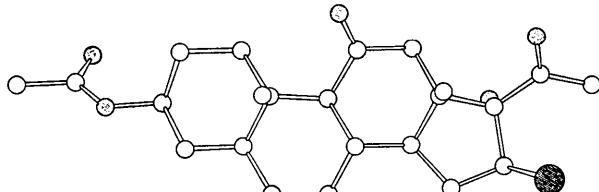
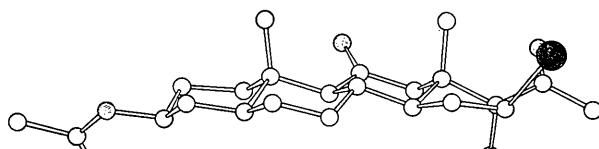


Fig. 3. Perspective views of the molecule. Large shaded circles: Br; small shaded circles: O; open circles: C. The plane of the acetate side chain lies at 55° to that of the *A* ring and that of the acetyl group at 40° to that of the *D* ring.

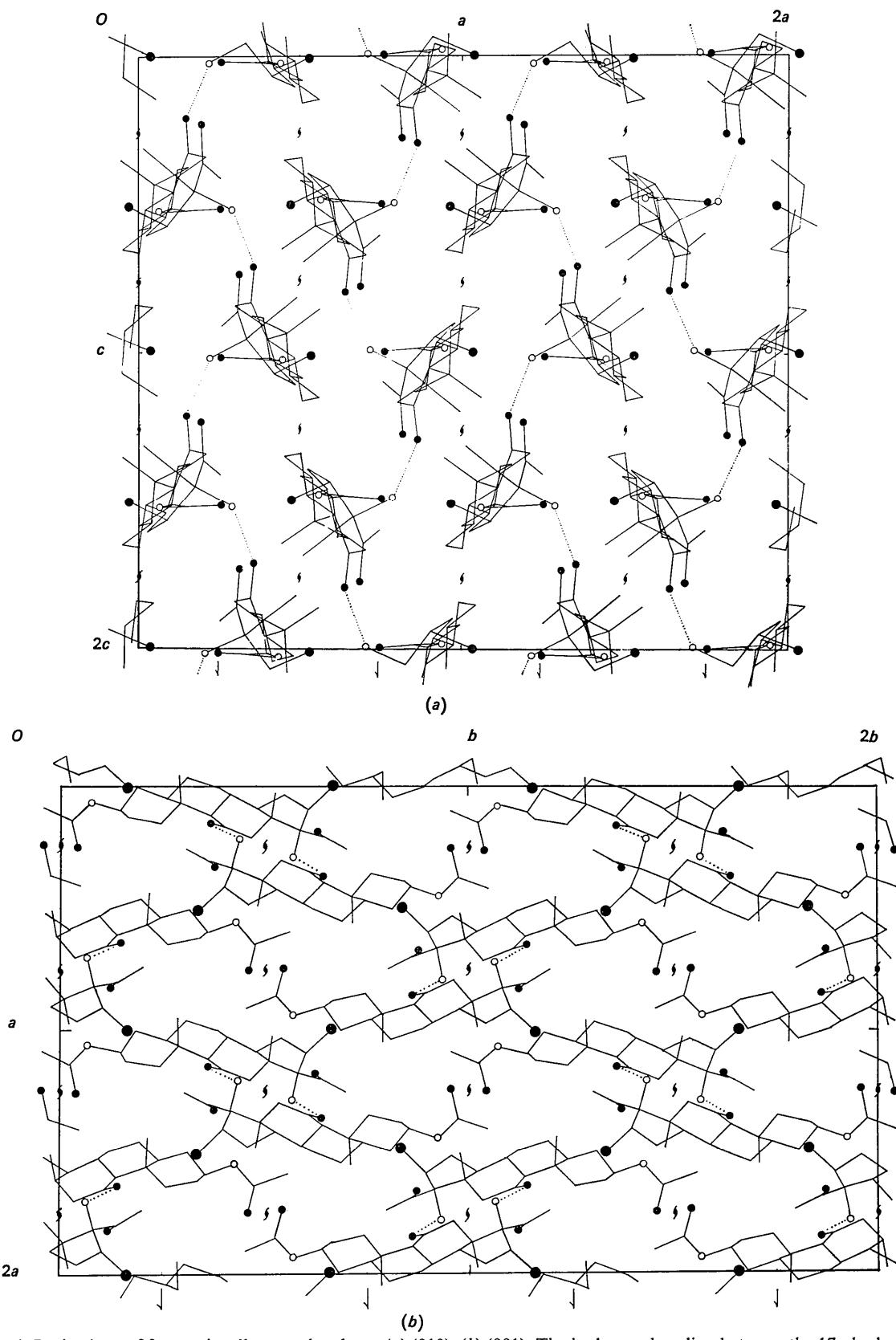


Fig. 4. Projections of four unit cells onto the planes (a) (010), (b) (001). The hydrogen bonding between the 17α -hydroxyl and the 11-keto oxygen is shown by dotted lines.

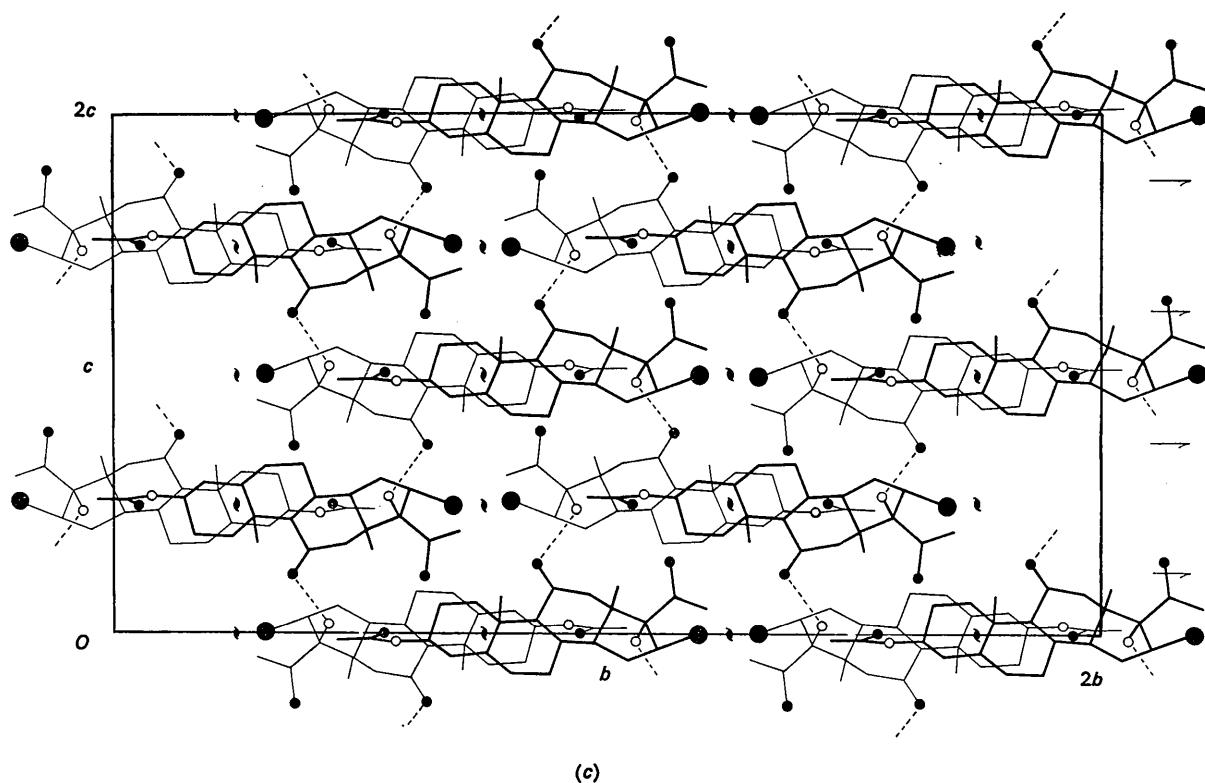


Fig.4 (cont.) (c) Projection of four unit cells onto the plane (100).

Table 3. Intermolecular distances less than 4.0 Å

The equivalent positions are:

$$\begin{array}{ccccc} 1 & x & y & z \\ 2 & \frac{1}{2}-x & \bar{y} & \frac{1}{2}+z \\ 3 & \bar{x} & \frac{1}{2}-y & \frac{1}{2}-z \end{array}$$

O(20)-C(14) (2/01 $\bar{1}$) means O(20) at equivalent position 1 to [C(14)] at equivalent position 2, translated one unit cell in the y direction and one unit cell in the \bar{z} direction].

	Position	Distance		Position	Distance
O(11)-O(17)	2/01 $\bar{1}$	2.70 Å	C(12)-C(15)	2/01 $\bar{1}$	3.94 Å
O(11)-C(16)	2/01 $\bar{1}$	3.48	C(12)-C(16)	2/01 $\bar{1}$	3.92
O(11)-C(17)	2/01 $\bar{1}$	3.63	C(18)-O(3)	3/1 $\bar{1}$ 1	3.86
O(11)-C(21)	2/01 $\bar{1}$	3.76	O(20)-C(2)	3/1 $\bar{1}$ 1	3.82
C(11)-O(17)	2/01 $\bar{1}$	3.50	O(20)-C(19)	3/1 $\bar{1}$ 1	3.63
C(12)-O(17)	2/01 $\bar{1}$	3.49	C(18)-C(2)	3/1 $\bar{1}$ 1	3.80
O(20)-C(7)	2/01 $\bar{1}$	3.70	C(18)-C(22)	3/1 $\bar{1}$ 1	3.99
O(20)-C(9)	2/01 $\bar{1}$	3.81	C(18)-C(23)	3/1 $\bar{1}$ 1	3.96
O(20)-C(14)	2/01 $\bar{1}$	3.94	C(21)-C(19)	3/1 $\bar{1}$ 1	3.92

The acetate side chain [C(3), O(3), C(22), O(22), and C(23)] is planar but this plane lies at 55° to the plane of the *A* ring, instead of being perpendicular to it as might be expected (Mathieson, 1965), (see Fig. 3). This effect is due to the steric hindrance between the acetate

group of one molecule and the C(18) methyl group of adjacent molecules, which approach each other to within 3.8 Å. The acetyl side chain is planar, lying at 40° to the plane of the *D* ring and C(13) lies in this plane.

Table 4. Comparison between the values of $|F_0|$ and $|F_c|$ calculated for the refined atomic parameters
 $|F_0|$ and $|F_c|$ are given on ten times the absolute scale.

H	K	L	FCBS	FCAL	H	K	L	FCBS	FCAL	H	K	L	FCBS	FCAL	H	K	L	FCBS	FCAL	H	K	L	FCBS	FCAL						
0	0	2	2322	2849	C	11	7	121	207	1	2	11	9C	48	1	12	1C	78	1C1	2	10	97	M1	2	13	2	162	150		
0	0	4	856	98	C	12	2	274	296	1	2	12	4C	48	1	13	9C	96	2C	2	11	21	248	2	13	8	263	249		
0	0	6	525	656	C	11	9	93	66	1	3	0	303	322	1	13	1	16C	141	2	12	73	1C	2	13	4	126	179		
0	0	8	285	330	C	11	10	93	10	1	3	1	156	1150	1	13	2	3	0	81	200	2	13	1	5	283	310			
0	0	12	157	193	C	11	11	93	10	1	3	1	156	1150	1	13	1	159	120	2	12	73	1C	2	13	4	126	178		
0	0	1	276	1C2	C	12	0	1C5	1C7	1	3	1	1017	1224	1	13	4	145	13M	2	12	3	153	146	2	13	5	156	193	
0	1	2	400	461	C	12	1	27	1	3	4	291	252	1	13	5	187	197	2	13	3	3	656	63C	2	13	9	91	135	
0	1	3	1115	1059	C	12	2	633	673	1	3	5	60C	721	1	13	6	98	35	2	13	4	293	256	2	13	9	45	48	
0	1	4	93	86	C	12	3	85	42	1	3	6	197	237	1	13	7	146	149	2	12	3	503	479	2	13	4	405	408	
0	1	5	626	666	C	12	4	108	3C5	1	3	6	456	460	1	13	7	147	151	2	12	3	504	309	2	13	4	155	110	
0	1	6	75	82	C	12	5	10	10	1	3	6	169	209	1	13	7	148	150	2	12	3	504	305	2	13	4	158	179	
0	1	7	262	231	C	12	6	16	148	1	3	7	98	96	1	13	7	151	152	2	12	3	8	18	159	2	14	4	215	221
0	1	8	156	158	C	12	7	29	96	1	3	10	147	124	1	14	2	148	96	2	14	4	248	258	3	14	5	233	242	
0	1	9	262	251	C	12	8	97	1CC	1	3	11	9C	33	1	14	2	223	232	2	13	10	98	48	3	14	5	142	124	
0	1	10	138	1C3	C	12	9	91	53	1	3	12	74	22	1	14	4	162	174	2	13	11	84	48	3	14	6	9	95	
0	1	11	11	1C1	C	12	10	1C1	115	1	4	13	131	151	1	14	5	148	150	2	14	6	99	100	3	14	6	9	96	
0	1	12	13	1C1	C	12	11	1C1	115	1	4	14	131	151	1	14	5	151	153	2	14	6	100	80	3	14	6	9	97	
0	2	1	9d	4C	C	13	1	316	114	1	4	1006	993	1	14	6	99	36	2	14	4	454	414	2	14	5	80	49		
0	2	2	157	164	C	13	2	6	874	900	1	4	6	92	49	2	14	3	653	634	2	14	10	65	73	3	14	5	1	86
0	2	3	415	356	C	13	3	362	364	1	4	6	793	287	1	14	3	254	289	2	14	3	430	337	3	14	5	2	13C	
0	2	4	372	359	C	13	4	193	212	1	4	6	291	354	1	14	3	281	311	2	14	5	348	370	3	14	6	3	141	
0	2	5	400	384	C	13	5	6	67	1	4	6	456	460	1	14	3	214	247	2	14	5	152	154	3	14	6	1	143	
0	2	6	139	141	C	13	6	7	67	1	4	6	333	354	1	14	3	219	289	2	14	5	200	231	3	14	6	295	297	
0	2	7	267	3C9	C	13	7	28	14	1	4	6	32C	349	1	14	3	16C	154	2	14	5	7	205	229	2	14	5	165	166
0	2	8	164	147	C	13	8	0	4C4	327	1	4	7	95	36	1	14	3	5	217	2	14	8	13C	98	2	14	5	200	205
0	2	9	97	96	C	14	1	117	73	1	4	10	174	160	1	15	6	131	118	2	14	9	158	168	2	15	6	97	128	
0	2	11	91	39	C	14	2	229	206	1	4	12	72	54	1	15	7	94	36	2	14	10	96	38	2	15	7	103	93	
0	2	12	127	134	C	14	3	131	182	1	4	13	72	54	1	15	7	101	81	2	14	11	87	81	2	15	7	102	97	
0	3	1	311	193	C	14	4	77	77	1	4	14	72	70	1	15	9	96	37	2	14	12	24	24	2	14	13	10	21	
0	3	2	254	255	C	14	5	97	71	1	4	15	192	4C5	1	15	6	57	77	2	14	5	521	595	2	14	6	404	354	
0	3	3	325	371	C	14	6	192	202	1	4	16	505	508	1	16	0	235	266	2	14	5	252	253	2	14	6	307	299	
0	3	4	430	324	C	14	7	84	77	1	4	17	56	449	1	16	1	150	156	2	14	5	984	992	2	14	6	353	530	
0	3	5	488	4H5	C	14	8	92	49	1	4	18	277	355	1	16	2	263	267	2	14	5	242	243	2	14	6	218	228	
0	3	6	414	426	C	14	9	62	49	1	4	19	277	355	1	16	2	263	267	2	14	5	242	243	2	14	6	218	228	
0	3	7	211	146	C	14	10	15	315	1	4	20	200	203	1	16	3	176	136	2	14	6	61	81	2	14	6	163	181	
0	3	8	93	6C	C	15	2	224	227	1	5	13	119	139	1	16	5	146	125	2	14	7	167	116	2	15	7	201	157	
0	3	9	9d	36	C	15	3	453	443	1	5	10	96	59	1	16	6	95	49	2	14	7	149	191	2	15	7	8	97	
0	3	10	97	74	C	15	4	329	371	1	5	11	87	38	1	16	7	91	43	2	14	10	95	111	2	15	7	9	98	
0	3	12	60	5C	C	15	5	208	205	1	5	12	7C	4	1	16	8	82	80	2	14	11	85	36	2	15	7	102	93	
0	3	13	64	6C	C	15	6	192	190	1	5	13	206	204	1	16	9	96	36	2	14	12	82	80	2	15	7	104	96	
0	3	14	253	222	C	15	7	9	70	1	5	14	59	355	1	16	9	96	66	2	14	13	275	276	2	15	7	105	104	
0	3	15	178	192	C	15	8	29	39	1	5	15	158	155	1	16	1	198	190	2	14	14	276	272	2	15	7	106	104	
0	3	16	4	155	C	15	9	121	1C5	1	5	16	262	265	1	16	2	174	209	2	14	15	311	308	2	15	7	122	130	
0	3	17	5C1	461	C	15	10	121	1C5	1	5	17	3	47	1	16	3	224	209	2	14	16	311	308	2	15	7	122	130	
0	3	18	232	217	C	15	11	121	1C5	1	5	18	326	316	1	16	4	125	122	2	14	17	326	316	2	15	7	122	130	
0	3	19	70	51	C	15	12	0	21	1	5	19	6C	69	1	16	2	174	171	2	14	18	326	316	2	15	7	122	130	
0	3	20	453	222	C	15	13	1d1	4C	1	5	20	62C	530	1	16	2	174	171	2	14	19	341	351	2	15	7	122	130	
0	3	21	19	10	C	15	14	19	1C1	1	5	21	74	105	1	16	3	203	91	2	14	20	347	339	2	15	7	122	130	
0	3	22	203	220	C	15	15	20	84	1	5	22	74	105	1	16	3	203	91	2	14	21	341	351	2	15	7	122	130	
0	3	23	99	1C9	C	15	16	2	92	1	5	23	74	105	1	16	3	203	91	2	14	22	341	351	2	15	7	122	130	
0	3	24	303	293	C	15	17	3	76	1	5	24	92	117	1	16	3	203	215	2	14	23	303	293	2	15	7	122	130	
0	3	25	197	201	C	15	18	4	69	1	5	25	45	459	1	16	3	203	215	2	14	24	303	293	2	15	7	122	130	
0	3	26	91	31	C	15	19	5	69	1	5	26	459	459	1	16	3	203	215	2	14	25	303	293	2	15	7	122	130	
0	3	27	117	134	C	15	20	6	105	1	5	27	459	4																

Table 4 (cont.)

H	K	L	F0BS	FCAL	H	K	L	F0BS	FCAL	H	K	L	F0BS	FCAL	H	K	L	F0BS	FCAL	H	K	L	F0BS	FCAL							
4	4	0	243	179	4	14	6	97	94	5	4	10	85	67	5	16	1	192	167	6	10	80	E0	7	0	1	79	48			
4	4	1	175	292	4	14	7	92	79	5	4	11	75	68	5	16	2	177	159	6	11	63	30	7	0	2	81	27			
4	4	3	176	151	4	14	9	72	32	5	5	1	225	222	5	14	5	165	180	6	7	4	166	451	7	0	5	92	43		
4	4	5	80	28	4	15	1	146	107	5	5	3	255	259	5	16	5	92	66	6	7	2	227	267	7	0	6	140	140		
4	4	6	180	146	4	15	3	182	190	5	5	4	145	109	5	16	7	78	112	6	7	4	257	297	7	0	11	62	39		
4	4	7	147	68	4	15	3	182	190	5	5	5	315	288	5	16	8	100	98	6	7	4	269	279	7	0	11	62	39		
4	4	8	97	118	4	15	5	97	64	5	5	6	256	294	5	17	0	97	126	6	7	6	202	217	7	1	2	78	78		
4	4	9	98	41	4	15	6	94	34	5	5	5	180	160	5	17	1	91	126	6	7	6	202	217	7	1	2	244	240		
4	4	10	2	54	4	15	7	99	52	5	5	8	167	150	5	17	2	96	113	6	7	8	8	99	7	1	2	223	241		
4	4	11	80	51	4	15	M	79	48	5	5	9	96	89	5	17	3	129	115	6	8	0	83	33	7	1	3	252	227		
4	4	12	62	22	4	15	2	66	79	5	5	10	20	70	5	17	4	92	115	6	8	1	203	189	7	1	4	69	68		
4	4	5	60	696	709	4	16	C	98	119	5	5	11	73	47	5	17	5	88	72	6	8	2	394	457	7	1	5	246	246	
4	4	5	1	677	670	4	16	1	99	74	5	5	6	0	20	5	17	6	81	56	6	8	3	212	215	7	1	6	1	99	
4	4	5	146	249	4	17	2	151	111	5	5	6	10	70	5	17	7	78	138	6	8	4	228	264	7	1	3	97	44		
4	4	5	34	14	4	17	3	99	41	5	5	6	476	484	5	18	6	158	19C	6	8	5	267	250	7	1	8	87	77		
4	4	5	211	210	4	16	4	97	97	5	5	6	310	278	5	18	1	94	92	6	8	6	134	134	7	1	9	91	134		
4	4	5	24	238	4	16	5	95	79	5	6	4	154	150	5	18	2	93	106	6	8	7	137	153	7	1	0	79	65		
4	4	5	6	274	314	4	16	6	91	92	5	6	5	163	152	5	18	3	92	103	6	8	8	65	59	7	1	1	97	55	
4	4	5	7	162	159	4	16	7	83	35	5	6	6	160	159	5	18	4	94	105	6	8	1	149	149	7	1	2	82	78	
4	4	5	23	23	4	17	8	92	52	5	5	9	97	89	5	17	5	124	115	6	8	10	230	228	7	1	3	94	94		
4	4	5	10	92	58	4	17	9	99	80	5	6	6	138	144	5	18	6	74	84	6	9	0	311	335	7	1	2	339	349	
4	4	5	11	79	30	4	17	1	99	31	5	5	9	94	78	5	17	7	63	24	6	9	1	140	132	7	2	3	376	376	
4	4	5	12	59	35	4	17	2	98	86	5	5	10	86	80	5	19	1	90	80	6	9	2	220	264	7	2	4	143	143	
4	4	6	0	319	344	4	17	3	97	87	5	5	6	10	277	5	17	5	93	19	6	9	3	222	212	7	2	5	274	266	
4	4	6	1	687	703	4	17	4	97	87	5	5	6	10	271	5	17	5	93	19	6	9	3	222	212	7	2	5	274	266	
4	4	6	5	165	165	4	17	5	92	79	5	5	7	2	236	205	5	18	5	95	134	6	9	3	134	134	7	1	2	242	242
4	4	6	3	563	540	4	17	0	88	31	5	5	7	3	153	126	5	18	0	95	106	6	9	0	144	131	7	1	2	141	141
4	4	6	4	194	209	4	17	7	77	45	5	5	7	4	205	220	5	18	0	102	104	6	9	0	131	131	7	1	2	141	141
4	4	6	5	275	303	4	17	8	65	46	5	5	7	5	160	119	5	18	1	94	133	6	9	0	131	131	7	1	2	141	141
4	4	6	6	265	246	4	18	0	97	61	5	6	7	6	175	142	5	18	2	82	59	6	9	0	98	75	7	1	2	141	141
4	4	6	7	195	218	4	18	1	97	61	5	6	7	6	176	142	5	18	2	82	59	6	9	0	98	75	7	1	2	141	141
4	4	6	8	21	23	4	18	2	96	106	5	6	7	6	174	204	5	18	2	82	59	6	9	0	98	75	7	1	2	141	141
4	4	6	9	97	123	4	18	3	94	113	5	5	9	7	40	52	5	18	2	193	167	6	9	1	140	132	7	2	3	115	115
4	4	6	10	92	96	4	18	4	138	151	5	5	7	10	85	75	5	18	2	191	175	6	9	1	140	132	7	2	3	115	115
4	4	6	11	76	70	4	18	5	87	118	5	5	7	11	61	66	5	18	2	192	173	6	9	1	140	132	7	2	3	115	115
4	4	7	1	658	702	4	18	6	79	50	5	5	8	0	381	368	5	18	3	68	31	6	9	0	142	142	7	1	2	141	141
4	4	7	2	325	296	4	18	7	69	48	5	5	8	0	381	368	5	18	3	68	31	6	9	0	142	142	7	1	2	141	141
4	4	7	3	205	233	4	18	8	69	48	5	5	8	2	292	302	5	18	2	94	59	6	9	0	142	142	7	1	2	141	141
4	4	7	4	267	242	4	19	9	126	117	5	5	8	3	128	100	5	18	2	92	62	6	9	0	142	142	7	1	2	141	141
4	4	7	5	87	329	4	19	2	92	92	5	5	8	4	222	219	5	18	2	92	62	6	9	0	142	142	7	1	2	141	141
4	4	7	6	147	162	4	19	3	128	132	5	5	8	5	196	208	5	18	0	142	147	6	9	0	142	147	7	1	2	141	141
4	4	7	7	146	154	4	19	4	86	86	5	5	8	6	145	145	5	18	0	142	146	6	9	0	142	146	7	1	2	141	141
4	4	7	8	99	110	4	19	5	121	148	5	5	8	7	148	150	5	18	0	142	148	6	9	0	142	148	7	1	2	141	141
4	4	7	9	97	143	4	19	6	94	96	5	5	8	8	245	247	5	18	0	142	143	6	9	0	142	143	7	1	2	141	141
4	4	7	10	92	91	4	19	7	99	77	5	5	8	2	245	247	5	18	0	142	143	6	9	0	142	143	7	1	2	141	141
4	4	7	11	70	21	4	19	8	91	41	5	5	8	2	245	247	5	18	0	142	143	6	9	0	142	143	7	1	2	141	141
4	4	7	12	87	87	4	19	9	87	53	5	5	8	2	325	325	5	18	0	142	143	6	9	0	142	143	7	1	2	141	141
4	4	7	13	92	93	4	19	10	82	53	5	5	8	2	325	325	5	18	0	142	143	6	9	0	142	143	7	1	2	141	141
4	4	7	14	93	93	4	19	11	93	59	5	5	8	2	325	325	5	18	0	142	143	6	9	0	142	143	7	1	2	141	141
4	4	7	15	91	93	4	19	12	92	59	5	5	8	2	325	325	5	18	0	142	143	6	9	0	142	143	7	1	2	141	141
4	4	7	16	92	93	4	19	13	92	59	5	5	8	2	325	325	5	18	0	142	143	6	9	0	142	143	7	1	2	141	141
4	4	7	17	92	93	4	19	14	92	59	5	5	8	2	325	325	5	18	0	142	143	6	9	0	142	143	7	1	2	141	141
4	4	7	18	92	93	4	19	15	92	59	5	5	8	2	32																

Table 4 (cont.)

H	K	L	FCHS	FCAL	H	K	L	FCHS	FCAL	H	K	L	FCHS	FCAL
9	0	5	93	66	9	15	0	126	120	10	12	2	85	41
9	0	6	97	52	9	15	1	no	35	10	12	3	85	101
9	0	8	87	28	9	15	2	115	117	10	12	5	74	99
9	1	0	212	171	9	15	3	no	46	10	12	6	65	45
9	1	2	149	141	9	15	4	no	46	10	12	7	52	45
9	1	3	146	141	9	15	5	68	11	10	12	8	125	12
9	1	4	149	131	9	16	6	no	17	10	13	4	76	83
9	1	5	95	89	9	16	7	no	11	10	13	5	81	33
9	1	6	91	62	9	16	8	no	11	10	13	6	56	48
9	1	7	91	62	9	16	9	no	11	10	13	7	12	1
9	1	8	86	29	9	16	10	no	11	10	14	1	81	58
9	1	9	74	24	9	16	11	no	71	10	14	2	75	53
9	2	1	263	242	9	17	0	no	74	10	14	3	75	83
9	2	2	93	65	9	17	1	no	74	10	14	4	83	22
9	2	3	208	272	9	17	2	no	76	10	14	5	72	22
9	2	4	98	112	9	17	3	no	76	10	14	6	52	21
9	2	5	168	164	9	17	4	no	76	10	14	7	52	20
9	2	6	97	56	9	17	5	no	76	10	14	8	57	20
9	2	7	131	127	9	17	6	no	76	10	14	9	57	20
9	2	8	85	30	10	0	7	no	40	10	16	0	6	16
9	2	9	70	57	10	0	8	no	40	10	16	1	57	16
9	2	10	246	272	10	0	9	no	40	10	16	2	57	16
9	2	11	145	117	10	1	0	no	41	10	16	3	79	121
9	2	12	234	240	10	1	1	no	41	10	16	4	62	60
9	2	13	262	235	10	1	2	no	41	10	16	5	62	53
9	2	14	140	140	10	1	3	no	41	10	16	6	80	80
9	2	15	169	165	10	1	4	no	41	10	16	7	124	124
9	2	16	96	64	10	1	5	no	41	10	16	8	59	59
9	2	17	92	51	10	1	6	no	41	10	16	9	163	167
9	2	18	84	96	10	1	7	no	41	10	16	10	89	71
9	2	19	71	24	10	2	0	no	42	10	16	11	57	45
9	2	20	321	299	10	2	1	no	225	10	16	1	68	49
9	2	21	91	42	10	2	2	no	53	10	16	2	66	53
9	2	22	188	242	10	2	3	no	264	10	16	3	28	89
9	2	23	148	140	10	2	4	no	24	10	16	4	58	51
9	2	24	95	95	10	2	5	no	25	10	16	5	60	56
9	2	25	99	118	10	2	6	no	43	10	16	6	66	46
9	2	26	97	102	10	2	7	no	66	10	16	7	12	1
9	2	27	109	105	10	2	8	no	21	10	16	8	84	77
9	2	28	96	64	10	2	9	no	83	10	16	9	77	59
9	2	29	124	141	10	2	10	no	142	10	16	10	23	12
9	2	30	83	70	10	2	11	no	94	10	16	11	74	59
9	2	31	245	245	10	2	12	no	245	10	16	12	57	45
9	2	32	174	220	10	2	13	no	220	10	16	13	67	59
9	2	33	386	386	10	2	14	no	386	10	16	14	57	46
9	2	34	274	275	10	2	15	no	274	10	16	15	57	46
9	2	35	234	240	10	2	16	no	234	10	16	16	57	46
9	2	36	262	235	10	2	17	no	262	10	16	17	57	46
9	2	37	140	140	10	2	18	no	140	10	16	18	57	46
9	2	38	176	176	10	2	19	no	176	10	16	19	57	46
9	2	39	113	113	10	2	20	no	113	10	16	20	57	46
9	2	40	84	84	10	2	21	no	84	10	16	21	57	46
9	2	41	169	169	10	2	22	no	169	10	16	22	57	46
9	2	42	176	176	10	2	23	no	176	10	16	23	57	46
9	2	43	91	49	10	2	24	no	91	10	16	24	57	46
9	2	44	82	58	10	3	0	no	48	11	3	2	147	151
9	2	45	68	25	10	3	1	no	102	11	3	3	105	105
9	2	46	152	152	10	3	2	no	102	11	3	4	105	105
9	2	47	163	163	10	3	3	no	102	11	3	5	105	105
9	2	48	97	96	10	3	4	no	57	11	3	6	105	105
9	2	49	6	6	10	3	5	no	29	11	3	7	105	105
9	2	50	93	64	10	3	6	no	21	11	3	8	105	105
9	2	51	96	64	10	3	7	no	49	11	3	9	105	105
9	2	52	97	64	10	3	8	no	49	11	3	10	105	105
9	2	53	89	47	10	3	9	no	45	11	3	11	105	105
9	2	54	79	40	10	4	0	no	102	11	4	1	92	61
9	2	55	66	48	10	4	1	no	60	11	4	2	105	105
9	2	56	6	6	10	4	2	no	55	11	4	3	105	105
9	2	57	7	6	10	4	3	no	55	11	4	4	105	105
9	2	58	6	6	10	4	4	no	55	11	4	5	105	105
9	2	59	6	6	10	4	5	no	55	11	4	6	105	105
9	2	60	6	6	10	4	6	no	55	11	4	7	105	105
9	2	61	6	6	10	4	7	no	55	11	4	8	105	105
9	2	62	6	6	10	4	8	no	55	11	4	9	105	105
9	2	63	6	6	10	4	9	no	55	11	4	10	105	105
9	2	64	6	6	10	4	10	no	55	11	4	11	105	105
9	2	65	6	6	10	4	11	no	55	11	4	12	105	105
9	2	66	6	6	10	4	12	no	55	11	4	13	105	105
9	2	67	6	6	10	4	13	no	55	11	4	14	105	105
9	2	68	6	6	10	4	14	no	55	11	4	15	105	105
9	2	69	6	6	10	4	15	no	55	11	4	16	105	105
9	2	70	6	6	10	4	16	no	55	11	4	17	105	105
9	2	71	6	6	10	4	17	no	55	11	4	18	105	105
9	2	72	6	6	10	4	18	no	55	11	4	19	105	105
9	2	73	6	6	10	4	19	no	55	11	4	20	105	105
9	2	74	6	6	10	4	20	no	55	11	4	21	105	105
9	2	75	6	6	10	4	21	no	55	11	4	22	105	105
9	2	76	6	6	10	4	22	no	55	11	4	23	105	105
9	2	77	6	6	10	4	23	no	55	11	4	24	105	105
9	2	78	6	6	10	4	24	no	55	11	4	25	105	105
9	2	79	6	6	10	4	25	no	55	11	4	26	105	105
9	2	80	6	6	10	4	26	no	55	11	4	27	105	105
9	2	81	6	6	10	4	27	no	55	11	4	28	105	105
9	2	82	6	6	10	4	28	no	55	11	4	29	105	105
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9	2	84	6	6	10	4	30	no	55	11	4	31	105	105
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9	2	86	6	6	10	4	32	no	55	11	4	33	105	105
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9	2	90	6	6	10	4	36	no	55	11	4	37	105	105
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9	2	94	6	6	10	4	40	no	55	11	4	41	105	105
9	2	95	6	6	10	4	41	no	55	11	4	42	105	105
9	2	96	6	6	10	4	42	no	55	11	4	43	105	105
9	2	97	6	6	10	4								